

10

CHEMICAL QUANTITIES

Chapter Test A

A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description on the line.

Column A

- _____ 1. standard temperature and pressure
- _____ 2. molar volume
- _____ 3. representative particle
- _____ 4. molar mass
- _____ 5. atomic mass
- _____ 6. Avogadro's number
- _____ 7. mole (mol)
- _____ 8. empirical formula
- _____ 9. percent composition
- _____ 10. molar mass

Column B

- a. the volume occupied by a mole of any gas at STP (22.4 L)
- b. the mass in amu of an atom of an element
- c. the number of representative particles in a mole of a substance
- d. the percent by mass of each element in a compound
- e. the mass of a mole of a substance
- f. the formula that gives the lowest whole-number ratio of the elements in a compound
- g. an atom, a formula unit, or a molecule
- h. the mass in grams of a mole of any pure substance
- i. 6.02×10^{23} representative particles of a substance
- j. 0°C and 101.3 kPa

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. How many hydrogen atoms are in 4 molecules of isopropyl alcohol, $\text{C}_3\text{H}_7\text{O}$?
- a. $4 \times (6.02 \times 10^{23})$
- b. 4
- c. 32
- d. $32 \times (6.02 \times 10^{23})$
- _____ 12. The mass of a mole of NaCl is its:
- a. molar mass.
- b. empirical formula.
- c. gram atomic mass.
- d. atomic mass.

Chapter 10 Chemical Quantities Quiz

Rudolf Benedikt



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Basic Chemical Principles Donada Peters, Edward I. Peters, 1988 **Code of Federal Regulations** , 1994
Addison-Wesley Chemistry , 1990 **Student Study Guide Fundamentals of Chemistry by David E. Goldberg**
Margaret G. Kimble, 1994 **Title 40 Protection of Environment Parts 790 to 999 (Revised as of July 1, 2013)** Office
of The Federal Register, Enhanced by IntraWEB, LLC, 2014-07 40 CFR Protection of Environment Study Guide/Selected
Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the
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Catalog , 1921 **Chemical Technology and Analysis of Oils, Fats and Waxes** Julius Lewkowitsch, 1922 **MATLAB**
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