

NAME \_\_\_\_\_

**FORMULAS AND NAMES OF MOLECULAR COMPOUNDS WORKSHEET – Answer Key**

**Introduction:** When you began chemistry class this year, you probably already knew that the chemical formula for carbon dioxide was  $\text{CO}_2$ . Today you will find out why  $\text{CO}_2$  is named that way. Naming chemical compounds correctly is of paramount importance. The slight difference between the names carbon monoxide ( $\text{CO}$ , a poisonous, deadly gas) and carbon dioxide ( $\text{CO}_2$ , a greenhouse gas that we exhale when we breathe out) can be the difference between life and death! In this activity, you will learn the naming system for molecular compounds.

**Instructions:** Answer the following questions on separate loose-leaf paper. Listen to the teacher for directions on which questions to work on. Work only on the questions specified by the teacher.

**PART 1: CHEMICAL FORMULAS AND NAMES FOR COVALENT COMPOUNDS**

**MODEL 1 – MOLECULAR COMPOUNDS**

Molecular Formula	Number of Atoms of First Element	Number of Atoms of Second Element	Name of Compound
$\text{ClF}$	1	1	Chlorine monofluoride
$\text{ClF}_5$	1	5	Chlorine pentafluoride
$\text{CO}$	1	1	Carbon monoxide
$\text{CO}_2$	1	2	Carbon dioxide
$\text{Cl}_2\text{O}$	2	1	Dichlorine monoxide
$\text{PCl}_5$	1	5	Phosphorus pentachloride
$\text{N}_2\text{O}_5$	2	5	Dinitrogen pentoxide

- Fill in the table to indicate the number of atoms of each type in the molecular formula. [6]  
**See Model 1**
- Examine the molecular formulas given in Model 1 for various molecular compounds. How many different *elements* are present in each compound shown? [1]  
**There are two elements present in each compound.**
- What type of compounds are represented by these formulas? Why? [3]  
**These are molecular compounds because there are two nonmetals in each formula.**
- Find all the compounds in Model 1 that have chlorine and fluorine in them. Explain why the name "chlorine fluoride" is not sufficient to identify a specific compound. [3]  
**There are two compounds with chlorine and fluorine. The name "chlorine fluoride" does not indicate which compound it refers to. It could be the compound with one fluorine atom or the compound with five fluorine atoms.**

# Chapter 10 Prentice Hall Chemistry Answers

**Alan Sherman, Sharon J. Sherman**



## Chapter 10 Prentice Hall Chemistry Answers:

*Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11  
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**Computational Chemistry** Errol G. Lewars, 2024-06-04 This is the fourth edition of the successful textbook on computational chemistry which continues to provide a comprehensive introduction to the theory and practice of computational chemistry Notable updates include a review of references up to mid 2023 encompassing recent developments in scientific journals books and software The evolving prominence of density functional theory DFT is emphasized and attention is given to the increasing application of artificial intelligence in computational chemistry The book maintains key features from the previous edition delving into the mathematical intricacies of ab initio and density functional methods at an introductory level Clear explanations of matrix methods are provided offering a direct approach to obtaining energy levels and molecular orbitals Additionally each chapter includes sets of Easier and Harder drill questions with suggested answers at the end of the book enhancing the learning experience The book is intended for upper year undergraduate and graduate students studying computational and theoretical chemistry and for self study by researchers in universities and industry to whom computational chemistry may be useful

*Prentice Hall Chemistry*, 2000

**Prentice Hall Grammar** Jeanette Adkins, 2003

**Soil and Groundwater Remediation** Chunlong Zhang, 2019-10-30 An introduction to the principles and practices of soil and groundwater remediation Soil and Groundwater Remediation offers a comprehensive and up to date review of the principles practices and concepts of sustainability of soil and groundwater remediation The book starts with an overview of the importance of groundwater resource quality contaminant sources types and the scope of soil and groundwater remediation It then provides the essential components of soil and groundwater remediation with easy to understand design equations calculations and the practical applications The book contains information on remediation basics such as subsurface chemical behaviors soil and groundwater hydrology and characterization regulations cost analysis and risk assessment The author explores various conventional and innovative remediation technologies including pump and treat soil vapor extraction bioremediation incineration thermally enhanced techniques soil washing flushing and permeable reactive barriers The book also examines the modeling of groundwater flow and contaminant transport in saturated and unsaturated zones This important book Presents the current challenges of remediation practices Includes up to date information about the low cost risk based sustainable remediation practices as well as institutional control and management Offers a balanced mix of the principles practices and sustainable concepts in soil and groundwater remediation Contains learning objectives discussions of key theories and example problems Provides

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**Chemistry, an Experimental Science** Chemical Education Material Study, 1963

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**Chemistry and Our Changing World** Alan Sherman, Sharon J. Sherman, 1992 Now updated and improved this book's unique combination of complete topical coverage easy to understand explanations and focus on problem solving will give you an appreciation of the excitement of chemistry and will show you how chemistry is used in the everyday world

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**Paleoclimatology** Colin P. Summerhayes, 2020-06-22 Life on our planet depends upon having a climate that changes within narrow limits not too hot for the oceans to boil away nor too cold for the planet to freeze over Over the past billion years Earth's average temperature has stayed close to 14-15 C oscillating between warm greenhouse states and cold icehouse states We live with variation but a variation with limits Paleoclimatology is the science of understanding and explaining those variations those limits and the forces that control them Without that understanding we will not be able to foresee future change accurately as our population grows Our impact on the planet is now equal to a geological force such that many geologists now see us as living in a new geological era the Anthropocene Paleoclimatology describes Earth's passage through the greenhouse and icehouse worlds of the past 800 million years including the glaciations of Snowball Earth in a world that was then free of land plants It describes the operation of the Earth's thermostat which keeps the planet fit for life and its control by interactions between greenhouse gases land plants chemical weathering continental motions volcanic activity orbital change and solar variability

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