

Chapter 17 Thermochemistry Study Guide

17.1 – 17.2 Thermochemical Equations

Honors Chemistry

1. Make the following conversions:
 - a. 444 cal to joules = 1.86×10^3 J
 - b. 1.8 kJ to joules = 1.8×10^3 J
 - c. 0.45 kJ to calories = 1.1×10^2 cal
2. Classify each of these processes as endothermic or exothermic:
 - a. condensing steam – **exo**
 - b. burning alcohol – **exo**
 - c. evaporating alcohol – **endo**
 - d. baking a potato – **endo**
3. The specific heat capacity for silver is 0.24 J/g°C. Calculate the energy required to raise the temperature of 150.0 g Ag from 273 K to 298 K. Calculate the molar heat capacity of silver.

$$\text{Energy: } q = (150.0)(.24)(298-273) = 9.0 \times 10^2 \text{ J}$$

$$\text{Molar heat capacity: } (0.24 \text{ J/g}^\circ\text{C})(107.87 \text{ g/mol Ag}) = 26 \text{ J/mol}$$

4. It takes 585 J of energy to raise the temperature of 125.6 g Hg from 20.0°C to 53.5°C. Calculate the specific heat capacity and the molar heat capacity of Hg.

$$\text{Specific heat capacity: } C = q/m\Delta t = 585/(125.6)(53.5-20) = 0.139 \text{ J/g}^\circ\text{C}$$

$$\text{Molar heat capacity: } (0.139 \text{ J/g}^\circ\text{C})(200.59 \text{ g/mol Hg}) = 27.9 \text{ J/mol}$$

5. A 46.2-g sample of copper is heated to 95.4°C and then placed in a calorimeter containing 75.0 g water at 19.6°C. The equilibrium temperature in the calorimeter is 21.8°C. Calculate the specific heat capacity of copper; assuming that all the heat lost by the copper is gained by the water.

$$C = 0.203 \text{ J/g}^\circ\text{C}$$

6. A 15.0-g sample of nickel metal is heated to 100.0°C and dropped into 55.0 g of water, initially at 23.0°C. Assuming that all the heat lost by the nickel is absorbed by the water; calculate the final temperature of the nickel and the water. The specific heat of nickel is 0.444 J/g°C.

$$\begin{aligned} -15.0(.444)(T_f - 100.0) &= 55.0(4.184)(T_f - 23.0) \\ -6.66T_f + 666 &= 230.12T_f - 5292.76 \\ 5958.76 &= 236.78 T_f \\ \mathbf{25.2^\circ\text{C}} &= T_f \end{aligned}$$

7. Chloe was running bath water and realized it was too hot. If she has 20.0 L of water in the tub at 95°C and then adds 15.8 L of water at 75°C, what will the final temperature of the water be?

$$\begin{aligned} -20(T_f - 95) &= 15.8(T_f - 75) \\ -20T_f + 1900 &= 15.8 T_f - 1185 \\ 3085 &= 35.8 T_f \end{aligned}$$

Ch17 Assessment Chemistry

Daniel F McAuley



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Prentice Hall Chemistry, 2000 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Chemical Mixtures and Combined Chemical and Nonchemical Stressors** Cynthia V. Rider, Jane Ellen Simmons, 2018-02-16 In this book both basic and advanced concepts are discussed for considering mixtures from initial exposure characterization through evaluation of risk associated with combined exposures This book will provide an introduction to key issues and multiple options for evaluating both the toxicity of mixtures as well as the risk associated with exposure to mixtures Additionally promising tools adapted from other disciplines will be discussed in the context of mixtures toxicology and risk assessment Finally the discussion will move beyond chemical mixtures to address incorporating non chemical stressors into toxicity studies and cumulative risk assessments Although exposure to multiple chemical and non chemical stressors is the rule not the exception consideration of mixtures in toxicology and risk assessment continues to be a significant challenge This book will be an essential resource for researchers and professionals in the fields of toxicology epidemiology exposure science risk assessment and statistics *Chemistry* John S. Phillips, Cheryl Wistrom, 2000 *The International Convention for the Prevention of Pollution from Ships* Henrik Ringbom, Aldo Chircop, 2025-02-12 This Commentary presents an in depth exploration of what is widely regarded as the most important International Maritime Organization IMO convention on vessel source pollution Leading international experts provide an authoritative analysis of the International Convention for the Prevention of Pollution from Ships 1973 78 and its subsidiary instruments collectively known as MARPOL [Cracking the GED Test with 2 Practice Tests, 2020 Edition](#) The Princeton Review, 2019-08-20 Make sure you re studying with the most up to date prep materials Look for the newest edition of this title Princeton Review GED Test Prep 2021 ISBN 9780525569398 on sale June 2020 Publisher s Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product [Concepts of Chemical Dependency](#) Harold E. Doweiko, 1990 This book is a comprehensive introduction to the nature of drugs including alcohol and nicotine and their effects on people Also included are methods of treatment for people who are dependent upon drugs **Princeton Review GED Test Prep 2021** The Princeton Review, 2020-06-02 A trusted reference for students taking the GED test provides content review that is 100% aligned with test objectives in language arts math science and social studies and includes more than 835 drill questions and two full length practice tests **Preventing Chemical Weapons** Michael Crowley, Malcolm Dando, Lijun Shang, 2018-08-20 The life and chemical sciences are in the

midst of a period of rapid and revolutionary transformation that will undoubtedly bring societal benefits but also have potentially malign applications notably in the development of chemical weapons Such concerns are exacerbated by the unstable international security environment and the changing nature of armed conflict which could fuel a desire by certain States to retain and use existing chemical weapons as well as increase State interest in creating new weapons whilst a broader range of actors may seek to employ diverse toxic chemicals as improvised weapons Stark indications of the multi faceted dangers we face can be seen in the chemical weapons attacks against civilians and combatants in Iraq and Syria and also in more targeted chemical assassination operations in Malaysia and the UK Using a multi disciplinary approach and drawing upon an international group of experts this book analyses current and likely near future advances in relevant science and technology assessing the risks of their misuse The book examines the current capabilities limitations and failures of the existing international arms control and disarmament architecture notably the Chemical Weapons Convention in preventing the development and use of chemical weapons Through the employment of a novel Holistic Arms Control methodology the authors also look beyond the bounds of such treaties to explore the full range of international law international agreements and regulatory mechanisms potentially applicable to weapons employing toxic chemical agents in order to develop recommendations for more effective routes to combat their proliferation and misuse A particular emphasis is given to the roles that chemical and life scientists health professionals and wider informed activist civil society can play in protecting the prohibition against poison and chemical weapons and in working with States to build effective and responsive measures to ensure that the rapid scientific and technological advances are safeguarded from hostile use and are instead employed for the benefit of us all

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Science ,2001 *Mine Pit Lakes* Devin N. Castendyk,L. Edmond Eary,2009 Water quality of pit lakes is one of the most critical environmental issues facing the global mining industry As ore grades decrease and operators strive to improve efficiency the number of active pit mines will continue to outpace their underground counterparts in the years ahead How will these water resources be protected for future generations while the mining industry continues to meet society s growing demands for raw materials The key to solving this dilemma is accurately predicting the water quality in advance of open pit mining That s the purpose of *Mine Pit Lakes* The third in a series of six handbooks by the Acid Drainage Technology Initiative Metal Mining Sector ADTI MMS this volume includes the latest thinking from dozens of internationally respected experts from Canada Germany Australia and the United States You ll learn both the theory and science of predicting pit lake water quality and get insights into the best practices of pit lake management This book is an indispensable resource for mining professionals and environmental regulators who are considering new open pit mines or are developing monitoring programs or closure strategies for existing ones **Dynamics Exposure and Hazard Assessment of Toxic Chemicals** Rizwanul Haque,1980 Ann Arbor Mich Ann Arbor Science c1980 **Risk Assessment of Chemicals in the Environment** Federation of European Chemical Societies,1988 Reviews the current status of risk assessment procedures as they relate to the environment *Hazard Assessment of Chemicals* Jitendra Saxena,1986-10-01 **Holt Chemistry** Ralph Thomas Myers,2004 *Environmental Toxicology and Chemistry* ,2003 **Civil Engineering Hydraulics Abstracts** ,1984

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