

Study Guide #1

Lecture #1 (Chapter 1 – What is Perception?)

- **Sensation vs. Perception (Historical and Contemporary Understandings)**
 - **Historical**
 - **Sensation:** transformation of physical energy into neural energy.
 - **Perception:** complex processes that involve higher-order mechanisms (interpretation and memory).
 - **Contemporary**
 - **Sensation:** the registering of a physical stimulus on our sensory receptors.
 - **Perception:** turning the sensory input into meaningful conscious experience, translation of the neural signal into usable information.
- **Transduction**
 - **Transduction:** change environmental energy to nerve impulses, how physical stimuli are converted into a sensory representation, the process of converting a physical stimulus into an electrochemical signal.
- **Stimulus**
 - **Stimulus:** an element of the world that impinges on our sensory systems.
- **Receptors**
 - **Receptors:** specialized cells that respond to environmental energy (e.g. visual pigments reacts to light), specialized neural cells that transduce a physical stimulus into an electrochemical signal, called a neural response.
- **Neural Response**
 - **Neural response:** the signal produced by receptor cells that can then be sent to the brain.
- **Action (As Defined in the Text)**
 - **Action:** any motor activity, such as moving one's eyes along a page or swinging a baseball bat at an incoming ball.
- **Phenomenology**
 - **Phenomenology:** our subjective experience of perception.
- **Aristotle Illusion**
 - **Aristotle illusion:** a single touch between the tips of two crossed fingers will be experienced as if there were two touches.
- **Motion Aftereffect and the Waterfall Illusion**
 - **Motion aftereffect:** a sensory experience that occurs after prolonged experience of visual motion in one particular direction.
- **Doctrine of Specific Nerve Energies**
 - **Doctrine of specific nerve energies:** argues that is it the specific neurons activated that determine the particular type of experience (e.g. activation of the optic nerve leads to visual experiences).
- **Hermann von Helmholtz and the Constructivist Approach**
 - **Hermann von Helmholtz:** a professor who was trained in medicine and physiology but became famous for his work in sensory perception and physics.
 - **Constructivism:** sensory signal itself is not enough to explain the richness of our experience. One must incorporate information from existing knowledge to

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IM Harris



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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1
 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
one mole of HCl would be fully neutralized by one mole of NaOH

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $H_2CO_3(aq) + 2NaOH(aq) \rightarrow Na_2CO_3(aq) + 2H_2O(l)$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction naoh aq c6h4 cooh cook aq c6h4 coona cook aq h2o l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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