

Biology Chapter 2 Study Guide

1. What is the basic unit of all matter?
2. Name the three subatomic particles and give the mass, charge and location in the atom of each one.
3. What is the atomic number?
4. What is the atomic mass?
5. Draw elements 1-10 on the Periodic Table using the example on page 35. Be sure to show the electrons in their proper energy levels. Hint: Your Periodic Table App gives this information.
6. Define isotope.
7. Since different isotopes of atoms exist, explain how this leads to an average atomic mass for each element.
8. The isotopes of Carbon are radioactive, which means they decay at a certain known rate. How can knowing the amount of each isotope of Carbon in a shell that you find at the beach be utilized?
9. What is a compound or molecule?
10. Why do atoms form compounds or bond?
11. Describe ionic bonds and give an example of an ionic compound.
12. Describe a covalent bond and give an example of a covalent compound.
13. Why is water such an important compound?
14. Draw a molecule of water.
15. A water molecule is polar due to the way the atoms in water bond together. What kind of bond can form between water molecules due to this polarity?
16. Describe the difference between a mixture, solution and suspension.
17. What is an acid and how do they form?
18. What is a base and how does it form?
19. Write the chemical equation to show the disassociation of water in a solution to hydrogen and hydroxide ions.
20. Draw the pH scale. Show the values that represent a base, an acid and neutral solution and give one example of each.
21. Name the six most important atoms found in living things.
22. Why is carbon such an important atom in organic molecules? Discuss how it bonds and describe the types of molecules it can form.
23. How many covalent bonds are there between the carbon atoms in acetylene?
24. List the four major carbon containing compounds found in living things.
25. What are carbohydrates and why do living things need them? Distinguish between monosaccharides, disaccharides and polysaccharides.
26. What is a chemical reaction?
27. Write a chemical equation showing the reaction that occurs between hydrogen and oxygen to form water. Label the products and reactants.

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**Eldon D.. Enger,Enger,Frederick C..
Ross**

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Table of Contents Biology Chapter 2 Study Guide

1. Understanding the eBook Biology Chapter 2 Study Guide
 - The Rise of Digital Reading Biology Chapter 2 Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology Chapter 2 Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology Chapter 2 Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology Chapter 2 Study Guide
 - Personalized Recommendations
 - Biology Chapter 2 Study Guide User Reviews and Ratings
 - Biology Chapter 2 Study Guide and Bestseller Lists
5. Accessing Biology Chapter 2 Study Guide Free and Paid eBooks
 - Biology Chapter 2 Study Guide Public Domain eBooks
 - Biology Chapter 2 Study Guide eBook Subscription Services
 - Biology Chapter 2 Study Guide Budget-Friendly Options

6. Navigating Biology Chapter 2 Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Biology Chapter 2 Study Guide Compatibility with Devices
 - Biology Chapter 2 Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology Chapter 2 Study Guide
 - Highlighting and Note-Taking Biology Chapter 2 Study Guide
 - Interactive Elements Biology Chapter 2 Study Guide
8. Staying Engaged with Biology Chapter 2 Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology Chapter 2 Study Guide
9. Balancing eBooks and Physical Books Biology Chapter 2 Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology Chapter 2 Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Biology Chapter 2 Study Guide
 - Setting Reading Goals Biology Chapter 2 Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biology Chapter 2 Study Guide
 - Fact-Checking eBook Content of Biology Chapter 2 Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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