

Q1:

Bond breaking is an endothermic change and bond forming is an exothermic change.

Bond energy is the amount of energy in kJ/mol needed to break one mole of the specified bond.



Use the following bond energies to determine whether this reaction is exothermic or endothermic. You must show your reasoning.

bond	bond energies in kJ/mol
C-Cl	338
C-H	412
Cl-Cl	242
H-Cl	431
C-C	348

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[3]

Q2:

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RJ Shavelson



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Children's Books in Print R R Bowker Publishing, Bowker, 1999-12 **Resources in Education**, 1996 *Prentice Hall Chemistry*, 2000 **El-Hi Textbooks & Serials in Print, 2005**, 2005 **Organic Chemistry** Francis A. Carey, 2000
A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless integration of molecular modeling exercises and state of the art multimedia tools the 7th edition of Organic Chemistry clearly offers the most up to date approach to the study of organic chemistry *The Australian Mathematics Teacher*, 1998 **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02
Machinery, 1920 **Kentucky Administrative Regulations Service**, 2001 *Abstracts of Studies in Education and Psychology* Pennsylvania State University. College of Education, 1940 **Biology** Gerhart Campbell, 2000-09
Engineering Economic Analysis Donald G. Newnan, Jerome P. Lavelle, Ted G. Eschenbach, 2000 Highly regarded by professors and students alike Engineering Economic Analysis Eighth Edition introduces the fundamental concepts of engineering economics Written for standard engineering economics courses this bestselling volume by Donald G Newnan Jerome P Lavelle and Ted G Eschenbach covers essential time value of money principles for engineering projects and isolates the problems and decisions engineers commonly face It also examines the tools necessary to properly analyze and solve those problems Revised in 2000 the eighth edition focuses on the use of spreadsheets teaching students to use the enormous capabilities of modern software rather than relying on spreadsheet templates The majority of the chapters conclude with sections designed to help students create spreadsheets based on the material covered in each chapter The book s organization gives professors the flexibility to omit spreadsheet instruction without loss of continuity accommodating shorter courses or to require that all computations be done with spreadsheets thus preparing students to use this essential tool for real life problems **Clinical Chemistry** Lawrence A. Kaplan, Amadeo J. Pesce, Steven C. Kazmierczak, 2003 This comprehensive up to date readable text acts as a complete clinical chemistry course and professional reference providing detailed specific information on the principles of clinical chemistry in laboratory diagnosis as well as the pathophysiologic changes that occur in disease and affect testing outcomes Explanations of Laboratory Techniques Part 1 lead the reader through various necessary laboratory techniques and practices Chapters on Pathophysiology Part 2 provide descriptions of how specific diseases affect the human body A companion CD ROM packaged with the book features Methods of Analysis a comprehensive Urinalysis Manual and an interactive Study Guide Workbook to reinforce concepts The book s clear writing

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