

Chemical Reactions of Copper: Key

Copper Recycling Lab

(Day 1)

Day 1 Data

1. Write the balanced chemical equation for the reaction between copper and nitric acid (HNO_3). For all reactants and products, include the state of matter.



Cancel spectators: NO_3^-



Copper Recycling Lab

(Day 2 & 3)

1. What evidence was there that a chemical reaction took place in conversion 1?

Formation of $\text{Cu(NO}_3)_2$ solution (blue), release of NO_2 (g) brown, heat change and condensation

2. Identify the type of chemical reaction for conversion 1.

Single Displacement

3. What was the limiting reactant in conversion 1 and how can you tell?

Cu (s) - All of the copper wire is used up

4. Classify the product in the flask after conversion 1 as a pure substance, homogeneous mixture, or heterogeneous mixture.

Homogeneous Mixture

5. Why is the copper (II) nitrate not visible once the reactions are done? Where did it go?

It is an aqueous solution - is dissolved

6. If the experiment stopped after conversion 1, what method could be used to recover this compound?

Evaporate off the water

Copper Post Lab Answers

Stephen DeMeo



Copper Post Lab Answers:

A Durability Test of Wood Posts in Hawaii-- Third Progress Report Roger G. Skolmen,1971 **Environmental Chemistry in the Lab** Ruth Ann Murphy,2022-08-31 Environmental Chemistry in the Lab presents a comprehensive approach to modern environmental chemistry laboratory instruction together with a complete experimental experience The laboratory experiments have an introduction for the students to read a pre lab for them to complete before coming to the lab a data sheet to complete during the lab and a post lab which would give them an opportunity to reinforce their understanding of the experiment completed Instructor resources include a list of all equipment and supplies needed for 24 students a lab preparation guide an answer key to all pre lab and post lab questions sample data for remote learners and a suggested rubric for grading the labs Additional features include Tested laboratory exercises with instructor resources for environmental science students Environmental calculations industrial regulation and environmental stewardship Classroom and remote exercises An excellent user friendly and thought provoking presentation which will appeal to students with little or no science background A qualitative approach to the chemistry behind many of our environmental issues today *Million Dollar Data: Building Confidence - Vol.1* Stephen DeMeo,2020-07-31 Global warming our current and greatest challenge is without precedent Among the many consequences that are impacting our society one unanticipated concern involves scientific truth When the President of the United States and others in his administration declare that global warming is fake science it calls into question what real science is and what real school science should be I will argue that real science is quality science one that is based on the rigorous collection of reliable and valid data To collect quality data requires bending over backwards to get things right and this is exactly what makes science so special Truth is made when scientists go this extra yard and devise controlled experiments collect large data sets confirm the data and rationally analyze their results Making scientific truth sounds difficult to do in the science laboratory but in reality there are many straightforward ways that truth can be constructed In the first of two volumes I discuss twelve such ways I call them Confidence Indicators that can allow students to strongly believe in their data and their subsequent results Many of these methods are intuitive and can be used by young students on the late elementary level all the way up to those taking introductory college science courses As in life science is not without doubt In the second volume I introduce the concept of scientific uncertainty and the indicators used to calculate its magnitude I will show that science is about connecting confidence with uncertainty in a specific manner what I refer to as the Confidence Uncertainty Continuum expression This important relationship epitomizes the scientific enterprise as a search for probabilistic rather than absolute truth This two volume set will contain a variety of ways that data quality can be instituted into a science curriculum To support its use many of the examples that I will present involve science teachers as well as student work and feedback from different grade levels and in different scientific disciplines Specific chapters will be devoted to reviewing the academic literature on data quality as well as describing my own personal research

on this important but often neglected topic [Science Up to Standards, Grades 5 - 8](#) Pam Walker, Elaine Wood, 1999-01-15 Science Up to Standards is loaded with activities based on the guidelines defined by the National Science Education Standards The format follows closely those standards which encourage teachers to provide students with activities requiring both partial and full inquiries In partial inquiry activities students are given some instructions on how to approach the problem solving issue In full inquiry activities students design their own strategies for finding answers This great resource provides activities in seven content areas Science as Inquiry Physical Science Life Science Earth and Space Science Science and Technology Science in Personal and Social Perspectives and History and Nature of Science **The Zinc and Iodine**

Book Stephen DeMeo, 2019-05-15 This book is for chemistry teachers who are thinking about reinventing their laboratory experiments that they provide to their students More than a collection of experiments it is an example of using a chemical theme to teach chemistry Instead of introducing many different chemicals per experiment as is the norm in most lab manuals this novel resource focuses on two commonly found elements Zinc and Iodine So what is so special about these elements At the heart of this resource is a colorful cyclic reaction between zinc and iodine one that produces a compound that can decompose back to its original elements This unique phenomenon demonstrates that matter not only changes but is also conserved through a chemical reaction Knowing that a compound can be the same but different than the reactants that formed it is to understand the essence of chemical change Complementing this reaction this book contains experimental activities that utilize the zinc and iodine theme to scaffold new concepts such as the properties of matter solid and gas stoichiometry equilibrium kinetics acid base chemistry and electrochemistry This teacher tested resource focuses on a set of safe substances that are appropriate for high school teachers who provide an advanced chemistry placement course and for college instructors teaching a first year chemistry laboratory sequence [Take-Home Physics: 65 High-Impact, Low-Cost Labs](#) Michael Horton, 2009-05-30

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science **Holt Earth Science**, 1994 **Using Multimedia Technology in Chemistry Pre-laboratory Preparation** Jeffrey Glen Yoder, 2002

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