

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

*Internal choices have been provided in one question each in Section B,
Section C and Section D.*

*All working, including rough work, should be done on the same sheet as, and adjacent to the rest
of the answer.*

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

*Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$
 $= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$*

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$, $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number = 6.023×10^{23} .

SECTION A – 14 MARKS

Question 1

(A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid,
high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Xtreme Chemistry Papers

Guillaume Favre



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1. Understanding the eBook Xtreme Chemistry Papers
 - The Rise of Digital Reading Xtreme Chemistry Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Xtreme Chemistry Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Xtreme Chemistry Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Xtreme Chemistry Papers
 - Personalized Recommendations

- Xtreme Chemistry Papers User Reviews and Ratings
- Xtreme Chemistry Papers and Bestseller Lists
- 5. Accessing Xtreme Chemistry Papers Free and Paid eBooks
 - Xtreme Chemistry Papers Public Domain eBooks
 - Xtreme Chemistry Papers eBook Subscription Services
 - Xtreme Chemistry Papers Budget-Friendly Options
- 6. Navigating Xtreme Chemistry Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Xtreme Chemistry Papers Compatibility with Devices
 - Xtreme Chemistry Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Xtreme Chemistry Papers
 - Highlighting and Note-Taking Xtreme Chemistry Papers
 - Interactive Elements Xtreme Chemistry Papers
- 8. Staying Engaged with Xtreme Chemistry Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Xtreme Chemistry Papers
- 9. Balancing eBooks and Physical Books Xtreme Chemistry Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Xtreme Chemistry Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Xtreme Chemistry Papers
 - Setting Reading Goals Xtreme Chemistry Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Xtreme Chemistry Papers
 - Fact-Checking eBook Content of Xtreme Chemistry Papers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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