

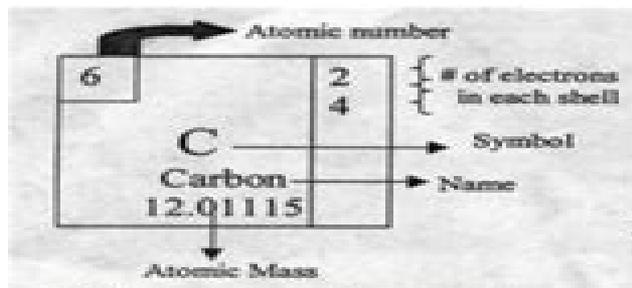
CHAPTER 19, SECTION 2
STUDENT NOTES

Atomic Mass

- The _____ contains most of the _____ of the atom because protons and neutrons are far more _____ than electrons.
- The _____ of a _____ is about the _____ as that of a _____— approximately
- The mass of each is approximately _____ times greater than the mass of the electron.

PROTONS IDENTIFY THE ELEMENT

- The number of _____ tells you what type of atom you have and vice versa. For example, every _____ atom has _____ protons. Also, all atoms with six protons are carbon atoms.
- The number of _____ in an atom is equal to a number called the _____.



3.0	Chlorine 17 Cl 35.45 3.0
2.5	

Using the information from the picture to the left, fill in the following blanks regarding the element:

- Name of Element: _____
 Atomic Mass: _____
 Atomic Number: _____
 Chemical Symbol: _____

Chapter 19 Section 2 Modern Chemistry

G Thomas



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