

CHAPTER 15 REVIEW

# Chemical Equilibrium

## Teacher Notes and Answers

### Chapter 15 Introduction 1

#### Question 1

- a.  $\frac{[CO_2]}{[CO]}$
- b.  $\frac{[CO_2]^2}{[CO]}$

- c.  $\frac{[CO_2]^2}{[CO]}$
- d.  $\frac{[CO_2]^2}{[CO]}$

e.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )

f.  $\frac{[CO_2]^2}{[CO]}$

g.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )

- h. correct
- i. correct

- 11. a. The sum of the stoichiometric coefficients of the reactants for reaction (a) is 2. The stoichiometric coefficients of the products and the other reactants are:

- b. 1
- c. 2

### Introduction 2

#### Question 1

- a.  $\frac{[CO_2]}{[CO]}$
- b.  $\frac{[CO_2]^2}{[CO]}$
- c.  $\frac{[CO_2]^2}{[CO]}$

- d. correct
- e. correct
- f. correct
- g. correct

- 11. These results are not included in equilibrium constants because they are not reversible reactions. The reaction  $2H_2 + O_2 \rightarrow 2H_2O$  is irreversible and so is

- 12. a. The sum of the stoichiometric coefficients of the reactants for reaction (a) is 2. The stoichiometric coefficients of the products and the other reactants are:

- b. 1
- c. 2

- d. correct
- e. correct

- 11. The stoichiometric coefficients of the reactants for reaction (a) are 2. The stoichiometric coefficients of the products and the other reactants are:
- b. 1
- c. 2

### 15.1 Equilibrium

#### Question 1

- a.  $\frac{[CO_2]}{[CO]}$
- b.  $\frac{[CO_2]^2}{[CO]}$
- c.  $\frac{[CO_2]^2}{[CO]}$
- d.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )
- e.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )
- f.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )
- g.  $\frac{[CO_2]^2}{[CO]}$  (using  $\nu$ )

- 11. The sum of the stoichiometric coefficients of the reactants for reaction (a) is 2. The stoichiometric coefficients of the products and the other reactants are:

- b. 1
- c. 2

# Chapter 18 Review Chemical Equilibrium Mixed Answers

**AW Rasmussen**



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