

4. Use particle theory to explain the results of your second test.  
Gas particles move around very quickly. When the gas is heated, the particles move even more quickly. This means they can take up a larger volume and so expand easily. Liquid particles are free to move around and slide over each other, but are not very free. When liquids are heated, the particles slide over each other more quickly but do not move apart as much as gas particles and so do not expand so easily.

### Investigating freezing (5-7)

In this activity, students melt a solid using hot water and then allow the solid to cool. As it does so, they measure the temperature at one-minute intervals. They plot their results on a graph and draw a line of best fit through the points, identifying the melting point from the cooling curve obtained.

Area	Equipment	Safety
<p>10.1 Show to outline how the particle theory of matter can be used to explain the properties of solids, liquids and gases, including changes of state.</p> <p>10.2 Choose appropriate apparatus and use it correctly.</p> <p>10.3 Make careful observations including measurements.</p> <p>10.4 Present results in the form of tables, bar charts or line graphs.</p>	<p>Each group will need:</p> <ul style="list-style-type: none"> <li>Practical worksheet (10)</li> <li>Large test tube/boiling tube (with stopper and side arm)</li> <li>Thermometer</li> <li>200 ml beaker</li> <li>Beaker or a similar bottle</li> <li>Clamp stand, boss and clamp</li> <li>Stopwatch</li> </ul>	<p>Eye protection must be worn at all times.</p> <p>Care must be taken when withdrawing the hot water into the beakers. The use of a kettle is to remove the need for heating up the water in the beaker using a Bunsen burner and the associated risk of spillage.</p> <p>During the heating of the solid the students should be encouraged to stand so that it splashes do occur they are ready to move out of the way.</p>

### Teaching notes

- Ask students to follow the instructions on the Practical worksheet (10).
- It is vital that the students have the boiling tube clamped in the correct position in the beaker and at the right height.
- It is often advisable to make sure that the water is almost boiling otherwise not all the solid will melt.
- Advise students how to anticipate when to take the reading (e.g. by counting down the last 10 seconds). They should also read the thermometer by adjusting their eye line to the height of the liquid in the thermometer.
- It is advisable to have more than one kettle per class because the water should be freshly boiled to be as hot as possible and the groups will probably need about 100–150 ml of boiling water to melt the solid.

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**Susan Fawcett**



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