

AP Chemistry: Chapter 17 Practice Test (Key at End)

All work must be done on separate paper or no credit will be earned. Some work or explanation must be shown for EACH problem or no credit will be earned for that question.

11. How many electrons are transferred in the following reaction?

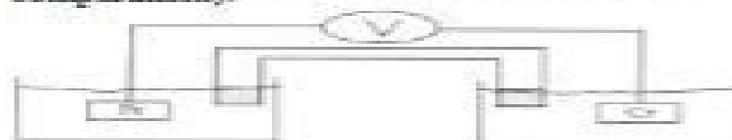


- [A] 3 [B] 4 [C] 6 [D] 10 [E] 2

19. What is the value of E for this cell at 25°C?

- [A] 1.88 V [B] 0.59 V [C] 1.76 V [D] 2.21 V [E] 2.12 V

Consider the galvanic cell shown below (the contents of each half-cell are written beneath each compartment):



0.50 M Br_2
0.10 M Br^-

0.20 M Cr^{3+}

The standard reduction potentials are as follows:



44. The standard free energies of formation of several aqueous species are:

	kJ/mol
$\text{H}^+(\text{aq})$	0
$\text{H}_2\text{O}(\text{l})$	-237
$\text{CH}_3\text{OH}(\text{aq})$	-163
$\text{HCOOH}(\text{aq})$	-351
e^-	0

What is the standard reduction potential of methanoic acid in aqueous solution (i.e., for $\text{HCOOH} + 4\text{H}^+ + 4\text{e}^- \rightarrow \text{CH}_3\text{OH} + \text{H}_2\text{O}$)?

- [A] -0.13 V [B] -0.25 V [C] +0.17 V [D] +0.25 V [E] +0.13 V

Chapter 17 Ap Chemistry

Sabine Zange



Chapter 17 Ap Chemistry:

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