

Introduction to Acids and Bases

1. Define acid and base using the Arrhenius and Bronsted-Lowry definitions.

	Arrhenius	Bronsted/Lowry
Acid		
Base		

2. Acids taste _____ and bases taste _____, bases also feel _____.
3. Acids turn litmus paper _____, bases turn litmus paper _____ and phenolphthalein _____.
4. Circle: Weak acids are strong / weak electrolytes. A weak electrolyte is a poor / good conductor of electricity.
5. Identify each of the following as an Arrhenius acid (A), Arrhenius base (B), or salt (S).
- i. NaCl _____ ii. KOH _____ iii. HCl _____ iv. Mg(OH)₂ _____
- v. MgCl₂ _____ vi. H₂SO₄ _____
6. List uses of acids in the industry.

7. List examples of common acids and bases found in a home or store.

Acids:

Base:

8. Label the Bronsted-Lowrey acid and base from the following reactions.





Acids And Bases Ws 4 Answer Key

David Baud



Acids And Bases Ws 4 Answer Key:

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