

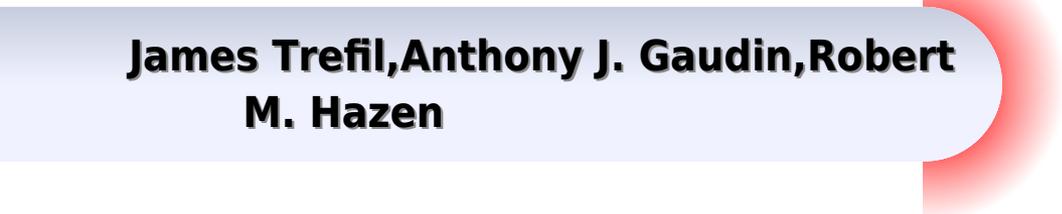
## UNIT 4: CHEMICAL REACTIONS STUDY GUIDE

Multiple choice:

- \_\_\_ **C** \_\_\_ 1. Chemical reactions \_\_\_\_\_.
- simply change the form or state of the reactants.
  - create new atoms.
  - always produce new substances.
  - create more matter than what they started with.
- \_\_\_ **B** \_\_\_ 2. Which of the following is the correct skeleton equation for the combination of solid zinc with liquid bromine to produce solid zinc bromide?
- $\text{Zn}_{2(s)} + \text{Br}_{(l)} \rightarrow \text{ZnBr}_{2(s)}$
  - $\text{Zn}_{(s)} + \text{Br}_{2(l)} \rightarrow \text{ZnBr}_{2(s)}$
  - $\text{Zn}_{(s)} + \text{Br}_{2(aq)} \rightarrow \text{ZnBr}_{2(s)}$
  - $\text{ZnBr}_{2(s)} \rightarrow \text{Zn}_{(s)} + \text{Br}_{2(l)}$
- \_\_\_ **A** \_\_\_ 3. A substance that is needed to speed up a reaction but is not a product or a reactant is known as a \_\_\_\_\_.
- catalyst
  - coefficient
  - spectator ion
  - steroid
- \_\_\_ **D** \_\_\_ 4. In a combustion reaction, one of the reactants is always \_\_\_\_\_.
- a liquid
  - water
  - a metal
  - oxygen
- \_\_\_ **B** \_\_\_ 5. Which of the following statements is true about the decomposition of a simple binary compound?
- The products cannot be predicted.
  - The reactant is a single substance.
  - The reactant must always be a molecular substance.
  - The product is a single substance.
- \_\_\_ **C** \_\_\_ 6. The reaction  $2\text{P}_2\text{O}_5 \rightarrow 4\text{P} + 5\text{O}_2$  is an example of which type of reaction?
- double replacement
  - single replacement
  - decomposition
  - combustion
  - synthesis

# Chemical Reaction Study Guide Answers

**James Trefil, Anthony J. Gaudin, Robert  
M. Hazen**



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*Study Guide, Chemical Principles, Fifth Edition, Zumdahl* Paul B. Kelter, Steven S. Zumdahl, 2004-04

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*Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry* Iris Stovall, Steven S. Zumdahl, 1996

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