

## Periodic Trends Webquest and Graphing

**Essential Question:** How do properties of atoms change across the periodic table?

Go to the following website: [www.ck12.org/periodic-table](http://www.ck12.org/periodic-table)

### Atomic Radius

- Using the information on the interactive periodic table, fill in the atomic radii for the following elements. You will have to click on each element, then click on atomic data. Use the atomic radius (non-bonded) in Å (angstroms).

#### A. Across a period (→)

Element	Atomic Number	# of Protons	# of Electrons	Atomic Radius
Be				
B				
C				
N				
O				
F				

Atomic Radius



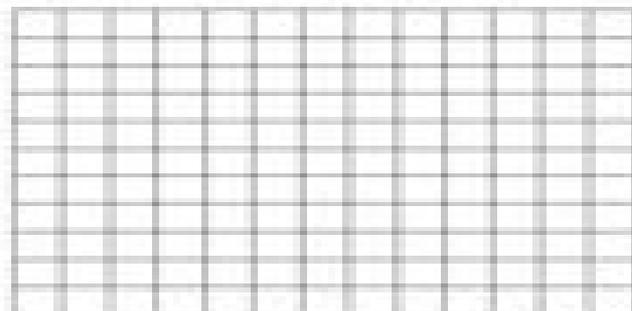
What is the trend for atomic radius across a period (→)?

Element

#### B. Down a group (↓)

Element	Atomic Number	# of Protons	# of Electrons	Atomic Radius
H				
Li				
Na				
K				
Rb				
Cs				

Atomic Radius



What is the trend for atomic radius down a group (↓)?

Element

# Chem Periodic Trends Webquest Answer Key

**SA Dillow**



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