

Paper 3

Questions are applicable for both core and extended candidates

- 1(a) Fig. 4.1 shows a gas syringe that contains 60 cm³ of ethene gas.



Fig. 4.1

State how the volume of ethene in the gas syringe changes when the temperature is decreased and the pressure remains the same.

..... [1]

- 2 Sulfur is an element in Group VI of the Periodic Table.

- (c) Sulfur is a solid at room temperature and pressure.

Describe the motion and separation of the particles in solid sulfur.

motion

.....

separation

.....

[2]

- (d) Liquid sulfur reacts with chlorine to produce disulfur dichloride.



- (i) Describe how the general physical properties of a liquid differ from those of a solid. Give **two** differences.

1

2

[2]

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