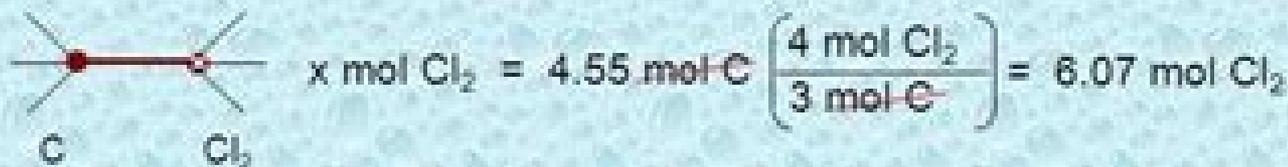


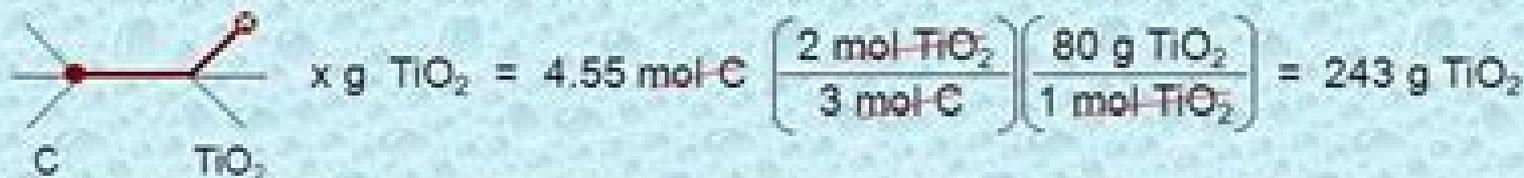
Stoichiometry



How many moles of chlorine will react with 4.55 moles of carbon?

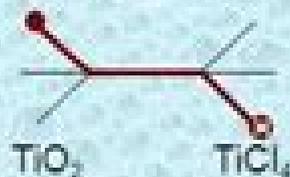


How many grams of titanium (IV) oxide will react with 4.55 moles of carbon?



How many molecules of TiCl₄ will react with 115 g TiO₂?

$$x \text{ molecules TiCl}_4 = 115 \text{ g TiO}_2 \left(\frac{1 \text{ mol TiO}_2}{80 \text{ g TiO}_2} \right) \left(\frac{2 \text{ mol TiCl}_4}{2 \text{ mol TiO}_2} \right) \left(\frac{6.02 \times 10^{23} \text{ molecules TiCl}_4}{1 \text{ mol TiCl}_4} \right)$$



$$= 8.66 \times 10^{23} \text{ molecules TiCl}_4$$

Basic Stoichiometry Pogil Answer Key

H Kauffman



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