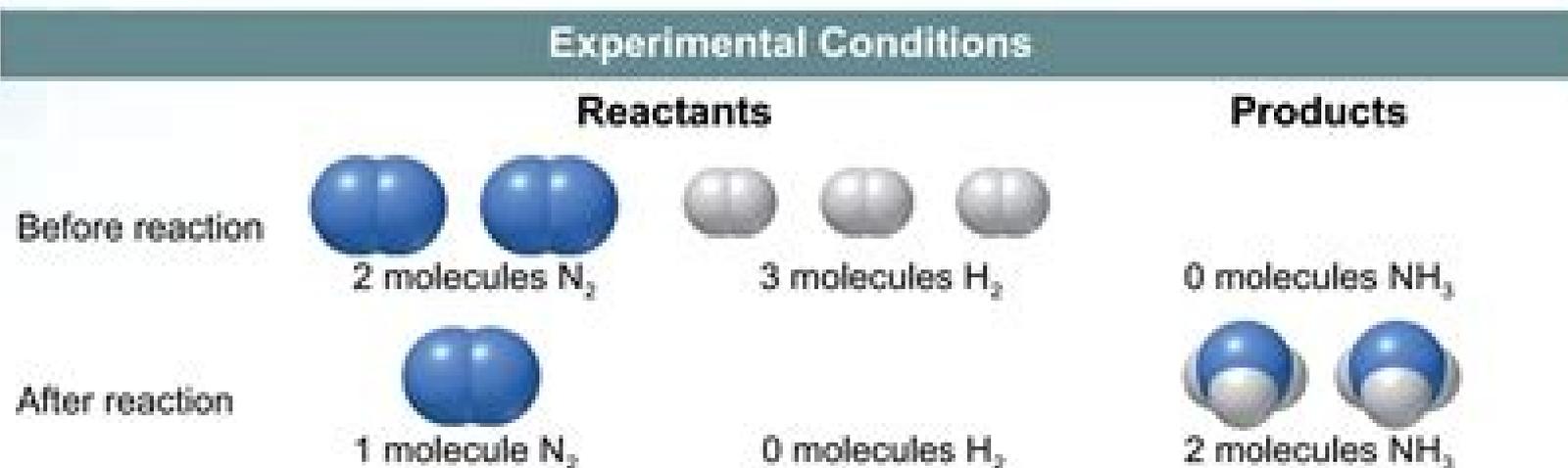


## 12.3 Limiting Reagent and Percent Yield >

## Limiting and Excess Reagents



- All the  $H_2$  has now been used up, and the reaction stops.
- One molecule (mole) of unreacted  $N_2$  is left in addition to the two molecules (moles) of  $NH_3$  that have been produced by the reaction.

## 9 3 Limiting Reagent And Percent Yield

**Siddappa N.Byrareddy**



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