

## Lab Report Format

Dry 4 – Qualitative Analysis of Cations

Charles A. Butler

### Part 1: Identification of Cations in Solution - $\text{Ag}^+$ , $\text{Hg}_2^{2+}$ , $\text{Pb}^{2+}$

#### 1. Introduction

- The objective of this experiment is to find which cations unknown solutions contain.
- We first start by obtaining our unknown solution. We add a NaCl to add chloride to our test tube. We then centrifuge it down. For our first cation we use heat to dissolve lead chloride. For our second test and second ion we add  $\text{NH}_3$  to produce either a soluble complex or a green/black solution that is our mercury. We then centrifuge and decant this and throw away the original test tube. We will now test the new test tube for the third ion. We do this by using PH 4. If there is silver it will precipitate into silver chloride.

#### 2. Results

##### Data Table

The ID number of your Chlorides Unknown: ___11___	
Possible Ions	Present (Y/N)
Ag	Yes
Hg	Yes
Pb	No

### Part 2: Identification of Cations in Solution - $\text{Co}^{2+}$ , $\text{Cr}^{3+}$ , $\text{Cu}^{2+}$

#### 3. Introduction

- The objective of this experiment is to find ions in an unknown solution.
- We first start by getting our unknown solution. Mine was ID 4. We mix in NaOH to see if we get a green mixture. Indicating Cobalt and Copper. We then centrifuge and Decant. We then take the decant test tube and inspect for green indicating Chromium. We add PH 10 and nitric acid to confirm chromium. Next, we go back to the Cu/Co tube and add  $\text{NH}_3$  to see if we get a cloudy mixture. We then centrifuge and decant. If we have a blue color, then we can confirm cobalt. Now we take the blue tube and add nitric acid, and this will help us to confirm for copper. From my testing the tubes all stayed clear indicating that no ion's were present.

#### 4. Results

##### Data Table

The ID number of your Transition Metal (II) Unknown: ___4___	
Possible Ions	Present (Y/N)
Co	no
Cr	no
Cu	no

### Part 3: Identification of Cations in Solution - $\text{Ba}^{2+}$ , $\text{Sr}^{2+}$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$

#### 5. Introduction

# Cations And Ions Lab Report

**Scott C. Dulebohn**



## **Cations And Ions Lab Report:**

*Prune Production Manual* Richard P. Buchner,2012-05-16 Written in easy to read non technical language this manual is the perfect field application guide Inside you ll find the professionalism expertise and science based answers you ve come to expect from the University of California with contributions from more than 40 Cooperative Extension professionals UC faculty USDA scientists and highly skilled prune industry experts Chapters include An industry overview A detailed description of prune biology Information on understanding soils varieties irrigation and fertilization Pest management techniques A lesson on harvest and postharvest management The breadth of expertise and knowledge contained in the 320 pages of this manual along with the more than 300 photos and 56 color illustrations make this one of the most comprehensive prune production manuals in the world

**Pistachio Production Manual** Louise Ferguson,David Haviland,2016-12-02 The long awaited Pistachio Production Manual from the University of California is here The combined knowledge of 42 UC and industry experts and years of research and field trials are brought to fruition in this long awaited 321 page manual From an overview of the state of the industry to physiological disorders the 8 part manual covers everything you need to know Chapters cover topics including orchard design rootstocks and cultivars planting and training young trees weed insect mite and vertebrate management irrigation and salinity management disease management and physiological disorders including alternate bearing nut blanking and shell splitting Over 200 color photographs and 60 diagrams charts and tables illustrate key points The back cover includes a photographic guide to the developmental stages of the pistachio

**Walnut Production Manual** David E. Ramos,1997-10-01 **Annual Report - Agricultural Experiment Station** Alabama Agricultural Experiment Station,1906 **Annual Report of the Agricultural Experiment Station of the Alabama Polytechnic Institute** Alabama Polytechnic Institute. Agricultural Experiment Station,1919 Laboratory Exercises for Preparatory Chemistry Kathy Dodds Tyner,1994-06 Laboratory Exercises for Preparatory Chemistry is the perfect complement to a one semester preparatory chemistry laboratory course Tyner s manual emphasizes the application of chemistry and the principles of science to everyday life The labs are directly applicable to the real world and often contain supplemental assignments that illustrate an application

Citrus Production Manual Louise Ferguson,Elizabeth Elliot Grafton-Cardwell,2014-04-09 Citrus production is complex requiring a delicate balancing act during the growing season and lots of preparation This new manual covers the many steps in the process in a clear and accessible way This manual also details the latest horticultural and disease issues affecting citrus production From deciding scion variety and rootstock to establishing an orchard to managing production to postharvest handling you ll find it all here in a readable format Colorful photos and clear diagrams and illustrations guide you through important concepts Chapters cover History Botany and Physiology Orchard Establishment Pest and Disease Management Postharvest Handling

Experiment Station Record United States. Office of Experiment Stations,1917 **General, Organic, and Biological Chemistry Study Guide and**

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*TID.*, 1973     *Pacific Northwest Laboratory Annual Report to the DOE Assistant Secretary for Environment* Pacific Northwest Laboratory, 1979     Selected Water Resources Abstracts, 1970     **U.S. Government Research & Development Reports**, 1970     **Technical Note**, 1917     **Chemistry of Life** Peter P. Berlow, 1982     *Radioactive Waste Management* U.S. Atomic Energy Commission, 1973     Bibliography of Scientific and Industrial Reports, 1965-07

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