

16.4 Calculations Involving Colligative Properties

Sample Problem 16.7

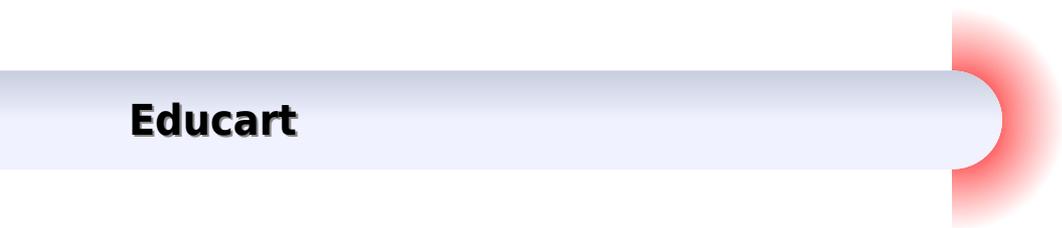
1 Analyze List the knowns and the unknown.

- According to the definition of molality, the final solution must contain 0.060 mol KI per 1000 g H_2O .
- Use the molality as a conversion factor to convert from mass of the solvent (H_2O) to moles of the solute (KI).
- Then use the molar mass of KI to convert from mol KI to g KI.

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Answer Sheet

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