

Name: _____

Grade and Section: _____

Module 1

What I Know

1	2	3	4	5
6	7	8	9	10
11	12	13	14	15

What's New

1	6
2	7
3	8
4	9
5	10

What is It

- 1
- 2
- 3
- 4
- 5

What's More

1	2	3	4	5
6	7	8	9	10

What I Have Learned

- 1
- 2
- 3
- 4
- 5

What I Can Do (use separate sheet)

Assessment

1	2	3	4	5
6	7	8	9	10
11	12	13	14	15

Module 2

What I Know

1	2	3	4	5
6	7	8	9	10
11	12	13	14	15

What's More

- a.
- b.
- c.

What I Have Learned (use a separate sheet)

What I Can Do (use a separate sheet)

Assessment

1	2	3	4	5
---	---	---	---	---

6	7	8	9	10
11	12	13	14	15

Module 3

What I Know

1	2	3	4	5
6	7	8	9	10
11	12	13	14	15

What's New (do the activity and take pictures)

a. Which of the following set-ups mixed well? Which did not?

b. What can you infer from the result of this activity?

Molecule	EN of the first atom	EN of the 2 nd atom	ΔEN
1. $NaCl$			
2. HCl			
3. Cl_2			
4. H_2			
5. H_2O			

What's More

Molecule	Lewis Structure	ΔEN	Bond Polarity	Molecular Geometry	Polarity of Molecule
1. PCl_3					
2. $BrCl_3$					
3. CH_4					
4. OP_3					
5. SP_3					

What I Have Learned

1

2

What I Can Do

1

2

233 Physical Science Answer Sheet

Bill W. Tillery



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