

CHEMISTRY PAPER-II

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

OBJECTIVE

Note: You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?
 (A) *Be* (B) *Ra* (C) *Ba* (D) *Rn*
- (2) Which element forms an ion with charge +3?
 (A) *Be* (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:
 (A) *N₂O* (B) *N₂O₃* (C) *N₂O₄* (D) *N₂O₅*
- (4) Chlorine heptaoxide (*Cl₂O₇*) reacts with water to form:
 (A) Hypochlorous acid (B) Chloric acid (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO₄* is:
 (A) *ClO₃* (B) *ClO₂* (C) *Cl₂O₅* (D) *Cl₂O₇*
- (6) Which of the following is a non-typical transition element?
 (A) *Cr* (B) *Mn* (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:
 (A) Position isomerism (B) Functional group isomerism (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:
 (A) Halogenation (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?
 (A) *AlCl₃* (B) *HNO₃* (C) *BeCl₂* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?
 (A) *E₁* and *E₂* (B) *E₂* and *S_N2* (C) *S_N1* and *E₂* (D) *E₁* and *S_N1*
- (11) Which compound is more soluble in water?
 (A) *C₂H₅OH* (B) *C₆H₅OH* (C) *CH₃COCH₃* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?
 (A) Methanal (B) Ethanal (C) Propanal (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?
 (A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?
 (A) $\frac{H_2}{Ni}$ (B) $\frac{H_2}{Pt}$ (C) *NaBH₄* (D) *LiAlH₄*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:
 (A) Alcoholic *KOH* (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO₃*
- (16) Most concentrated solid nitrogen fertilizer is:
 (A) *NH₃* (B) *(NH₄)₂HPO₄* (C) *(NH₂)₂CO* (D) *NH₄NO₃*
- (17) Mark the correct statement.
 (A) *Na⁺* is smaller than *Na* atom (B) *Na⁺* is larger than *Na* atom
 (C) *Cl⁻* is smaller than *Cl* atom (D) *Cl⁻* ion and *Cl* atom are equal in size

Chemistry Mcqs For Class 1chapter Wise

Gurukul



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Topic wise MCQs for NEET Chemistry: Environmental Chemistry, Organic Chemistry, P Block Elements, States of Matter, Redox Reaction, Chemical Equilibrium, Thermodynamics, Periodic Classification of Elements, s Block Elements, Chemical Bonding, Molecular Structure, Structure of Atom, Some Basic Concepts of Chemistry, Hydrocarbons, Some Basic Concepts of Chemistry, Solid State, Structure of Atom, Solutions, Classification of Elements and Periodicity in Properties, Electrochemistry, Chemical Bonding and Molecular Structure, Chemical Kinetics, States of Matter, Gases and Liquids, Surface Chemistry, Thermodynamics, General Principles and Processes of Isolation of Elements, Equilibrium, Redox Reactions, d and f Block Elements, Hydrogen, Coordination Compounds, Haloalkanes and Haloarenes, Some p Block Elements, Alcohols, Phenols and Ethers, Organic Chemistry, Some Basic Principles and Techniques, Aldehyde, Ketones and Carboxylic Acids, Hydrocarbons, Organic Compounds Containing Nitrogen, Environmental Chemistry, Biomolecules, Polymers and Chemistry in Everyday Life. Use a step wise approach to solve conceptual and complex questions. Several times Matching type Questions are asked where the students are required to find the mismatched or the correctly matched option. Some of the questions asked in the NEET exam are entirely memory based, therefore, the students are advised to memorize the common names, scientific names, concepts, and important definitions. Around 40% of the questions asked in the NEET exam are application based. Therefore, students need to focus more on the concepts along with its applications in order to score well in the examination. Chemistry topics broadly can be grouped into

Organic Inorganic and Physical chemistry Organic Chemistry is a vast section and holds the maximum weightage Its concepts and principles are important One needs to be conversant with properties of groups conversions recognizing organic compounds etc Interlinking between different chapters from organic chemistry can be of great help The students must primarily focus on reading NCERT textbooks Several times the questions asked in NEET exam are taken directly from the NCERT textbooks Initially avoid answering those questions for which you are not confident because your wrong answer may reduce your final score In order to utilize your time appropriately divide the three hours of examination time as per your comfort among Physics Chemistry and Biology Initially focus on attempting all easy questions and later on pick the difficult ones By this way your confidence will be elevated and you will also get more time to answer hard questions Practice previous years question papers mock tests and sample papers to get an idea on how to answer MCQ questions efficiently Preparing at an early stage is what an MCQ exam requires Avoid guesswork for negative marking questions as they might lower your final score These tips can be very helpful for students to answer difficult and brain teaser questions Prior preparations and practice are mandatory aspects of any examination Hence to crack highly competitive examination like NEET it is mandatory for students to prepare well and acquire the skills to tackle multiple choice questions effectively Rather than just following mere guesswork the aspirants can focus on the tips discussed to tackle Multiple Choice Questions in NEET in the right manner

Chemistry Quiz Umesh Kumar,2014-08-14 This book can be treated as a general chemistry quiz book for secondary level students It includes all fundamental chapters of chemistry For convenience of study it is divided into two parts Part 1 includes chapter pertaining to some elementary concepts of chemistry structure of atom atom and molecules and matter around us Part 2 includes chemical reactions and equations acid bases and salt metals and non metals carbon compounds and the last lesson deals with periodic classification of elements

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- 1 Study material strictly based on the Reduced Syllabus issued by the Board in July 2021 for Term 1 Exam
- 2 Focused on New Objective Paper Pattern Questions
- 3 Multiple Choice Questions MCQs based on the board s most recent typologies of the objective type questions a Stand Alone MCQs b Assertion Reason based questions c MCQs with a case study
- 4 Questions included from the official CBSE Question Bank issued in April 2021
- 5 NCERT NCERT Exemplar questions provided
- 6 2000 New Chapter wise Questions included for practice
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