

## Edgenuity Chemistry/Honors Chemistry B: Charles's Law Lab, U1L6

You may complete this worksheet OR do the lab in [Classkick](#). The class code is: LBD CCD. If you choose classkick, upload a document to the dropbox that lets me know it's completed!

All parts with **BOLD** lettering need to be answered!

Warm up:

**Why do you think a hot air balloon rises? What is the relationship between the temperature of a gas and its volume?**

In this lab, you will explore how the temperature of a gas affects the volume of that gas using a capillary tube, and then graphing that relationship.

[LINK TO VIRTUAL LAB](#)

Formulate an investigative question and hypothesis

**Question:** What is the effect of a gases  on it's volume?

**Hypothesis:**

**What is the independent variable?**

**What is the dependent variable?**

**Step 2: Measure the Capillary tube**

Using a ruler, measure the diameter of the capillary tube. Then divide by 2 to find the radius

**Radius of the capillary tube:**

**Step 3: Prepare the Capillary tube.** Follow the instructions from the virtual lab to attach a ruler to the capillary tube

**Step 4: Measure the volume of air near 0 degrees Celsius**

Place the capillary tube in the beaker. Record the temperature and the height of the gas column in the tube. To find the height, you will need to click on the capillary tube attached to the ruler. Make sure to read the height of the gas column from the bottom of the rubber stopper.

**Step 5-9: Measure the volume of air near 20, 40, 60, 80, and 100 degrees Celsius**

# Chemistry Edgenuity Unit 10

**Starline Press**



## **Chemistry Edgenuity Unit 10:**

**Chemistry Unit 10, 2nd Edition** Bright Thinker,2024-08      **Chemistry Skills Series** Frank L. Wiseman,1983-09-01

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