

SOLUTIONS

9. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass = 342 g mol⁻¹). The molar mass of solute 'X' is :
(a) 34.2 g mol⁻¹ (b) 57 g mol⁻¹
(c) 114 g mol⁻¹ (d) 3.42 g mol⁻¹ (1)
8. On mixing 10 mL of acetone with 50 mL of chloroform, the total volume of the solution will be :
(a) < 60 mL (b) > 60 mL
(c) = 60 mL (d) = 100 mL (1)
4. The system that forms maximum boiling azeotrope is :
(a) CS₂ and acetone (b) Benzene and toluene
(c) Ethanol and acetone (d) Chloroform and acetone (1)
5. Phenol dimerises in benzene having van't Hoff factor 0.54. Its degree of association is:
(a) 0.54 (b) 0.46
(c) 0.92 (d) 0.27
1. Low concentration of oxygen in the blood and tissues of people living at high altitudes is due to :
(a) Low temperature (b) Low atmospheric pressure
(c) High atmospheric pressure (d) High temperature (1)
4. For an electrolyte undergoing association in a solvent, the van't Hoff factor :
(a) is always greater than one
(b) has negative value
(c) has zero value
(d) is always less than one (1)
8. An azeotropic solution of two liquids has boiling point lower than either of them when it :
(a) is saturated
(b) shows positive deviation from Raoult's law
(c) shows negative deviation from Raoult's law
(d) shows no deviation from Raoult's law (1)
9. The relative lowering of vapour pressure of an aqueous solution containing non-volatile solute is 0.0225. The mole fraction of the non-volatile solute is :
(a) 0.80 (b) 0.725
(c) 0.15 (d) 0.0225 (1)
- 10*. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass = 342 g mol⁻¹). The molar mass of solute 'X' is :
(a) 34.2 g mol⁻¹ (b) 57 g mol⁻¹
(c) 114 g mol⁻¹ (d) 3.42 g mol⁻¹ (1)

Question numbers 13 to 16 are assertion reason types.

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David R. Klein



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