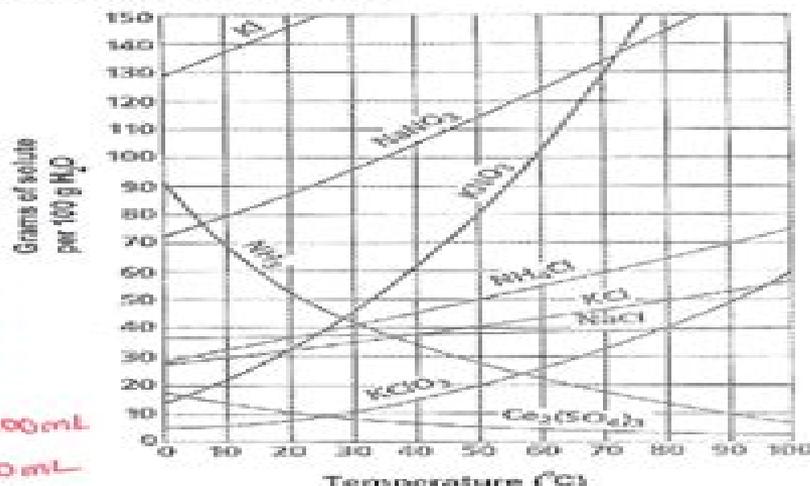


## Solubility Worksheet

### Honors Chemistry

1. Find the mass of solute that will dissolve in 100-mL of water at the following temperatures.

- a)  $\text{KNO}_3$  at  $70^\circ\text{C}$  = 130 g
- b)  $\text{NaCl}$  at  $100^\circ\text{C}$  = 40 g
- c)  $\text{NH}_4\text{Cl}$  at  $90^\circ\text{C}$  = 70 g
- d) Which of the above three substances is most soluble in water at  $15^\circ\text{C}$ . = NaCl



2. Determine the solubility of the following substances:

- a) What is the solubility of  $\text{KCl}$  at  $5^\circ\text{C}$ ? 28 g/100 mL
- b) What is the solubility of  $\text{KCl}$  at  $25^\circ\text{C}$ ? 34 g/100 mL
- c) What is the solubility of  $\text{Ce}_2(\text{SO}_4)_3$  at  $10^\circ\text{C}$ ? 13.5 g/100 mL
- d) What is the solubility of  $\text{Ce}_2(\text{SO}_4)_3$  at  $50^\circ\text{C}$ ? 5 g/100 mL

3. Solution	Unsaturated or Saturated? Explain	If unsaturated, how much more solute can dissolve in the solution?
a solution that contains 70g of $\text{NaNO}_3$ at $30^\circ\text{C}$ (in 100 mL $\text{H}_2\text{O}$ )	unsaturated	26 more grams
a solution that contains 50g of $\text{NH}_4\text{Cl}$ at $50^\circ\text{C}$ (in 100 mL $\text{H}_2\text{O}$ )	saturated	X
a solution that contains 20g of $\text{KClO}_3$ at $50^\circ\text{C}$ (in 50 mL $\text{H}_2\text{O}$ )	saturated	X
a solution that contains 70g of $\text{KI}$ at $0^\circ\text{C}$ (in 200 mL $\text{H}_2\text{O}$ )	unsaturated	190 more grams

4. At  $90^\circ\text{C}$ , you dissolved 10 g of  $\text{KCl}$  in 100. g of water. Is this solution saturated or unsaturated? unsaturated

5. A mass of 100 g of  $\text{NaNO}_3$  is dissolved in 100 g of water at  $80^\circ\text{C}$ .

- a) Is the solution saturated or unsaturated? unsaturated
- b) As the solution is cooled, at what temperature should solid first appear in the solution? Explain.

$35^\circ\text{C}$ , Solubility decreases as temperature decreases, when saturation occurs

6. Use the graph to answer the following two questions:

- Which compound is most soluble at  $20^\circ\text{C}$ ? KI
- Which is the least soluble at  $40^\circ\text{C}$ ?  $\text{Ce}_2(\text{SO}_4)_3$

# Chemistry Solubility Assessment Answer Key

**Léo Pomar**



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