

Lab: Specific Heat of a Metal

Suitable for Physical Science or Chemistry

Students in Grades 9 - 12.



Purpose of the Lab:

- Students will learn how to use a simple calorimeter.
- Students will calculate the specific heat of a metal using mass and temperature data obtained by using a calorimeter.
- Students will identify an unknown metal by determining its specific heat.

Materials List:

- 250 mL Beaker
- Hot plate or Bunsen burner
- Balance
- Styrofoam cup
- Thermometer
- Tongs
- Distilled water
- Copper strip
- Zinc strip
- Lead strip
- Unknown metal



Use this lab when teaching your units on physical properties or thermochemistry!

Chemistry Lab Specific Heat Answers

D Kirk



Chemistry Lab Specific Heat Answers:

Inquiry-based Experiments in Chemistry Valerie Ludwig Lechtanski, 2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student *Working with Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Laboratory Manual to Accompany Chemistry, [by] Stanley R. Radel, Marjorie H. Navidi Arthur D. Baker, 1990 *U Can: Chemistry I For Dummies* John T. Moore, Chris Hren, Peter J. Mikulecky, 2015-08-10 Now you can score higher in chemistry Every high school requires a course in chemistry for graduation and many universities require the course for majors in medicine engineering biology and various other sciences U Can Chemistry I For Dummies offers all the how to content you need to enhance your classroom learning simplify complicated topics and deepen your understanding of often intimidating course material Plus you ll find easy to follow examples and hundreds of practice problems as well as access to 1 001 additional Chemistry I practice problems online As more and more students enroll in chemistry courses the need for a trusted and accessible resource to aid in study has never been greater That s where U Can Chemistry I For Dummies comes in If you re struggling in the classroom this hands on friendly guide makes it easy to conquer chemistry Simplifies basic chemistry principles Clearly explains the concepts of matter and energy atoms and molecules and acids and bases Helps you tackle problems you may face in your Chemistry I course Combines how to with try it to form one perfect resource for chemistry students If you re confused by chemistry and want to increase your chances of scoring your very best at exam time U Can Chemistry I For Dummies shows you that you can *Chemistry* McGraw-Hill Staff, 2001-07

CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

Instructor's Guide for Introductory Chemistry in the Laboratory James F. Hall, 1996 **Science**, 1898 Vols for 1911 13 contain the Proceedings of the Helminthological Society of Washington ISSN 0018 0120 1st 15th meeting *Educational*

Data Mining Alejandro Peña-Ayala, 2013-11-08 This book is devoted to the Educational Data Mining arena. It highlights works that show relevant proposals, developments, and achievements that shape trends and inspire future research. After a rigorous revision process, sixteen manuscripts were accepted and organized into four parts as follows:

Profile The first part embraces three chapters oriented to:

- 1 describe the nature of educational data mining (EDM)
- 2 describe how to pre-process raw data to facilitate data mining (DM)
- 3 explain how EDM supports government policies to enhance education

Student modeling The second part contains five chapters concerned with:

- 4 explore the factors having an impact on the student's academic success
- 5 detect student's personality and behaviors in an educational game
- 6 predict students' performance to adjust content and strategies
- 7 identify students who will most benefit from tutor support
- 8 hypothesize the student answer correctness based on eye metrics and mouse click

Assessment The third part has four chapters related to:

- 9 analyze the coherence of student research proposals
- 10 automatically generate tests based on competences
- 11 recognize students' activities and visualize these activities for being presented to teachers
- 12 find the most dependent test items in students' response data

Trends The fourth part encompasses four chapters about how to:

- 13 mine text for assessing students' productions and supporting teachers
- 14 scan student comments by statistical and text mining techniques
- 15 sketch a social network analysis (SNA) to discover student behavior profiles and depict models about their collaboration
- 16 evaluate the structure of interactions between the students in social networks

This volume will be a source of interest to researchers, practitioners, professors, and postgraduate students aimed at updating their knowledge and find targets for future work in the field of educational data mining. *Basic*

Laboratory Principles in General Chemistry Fitzgerald B. Bramwell, 1990 **Lab Experiments for Modern Chemistry**
 Tzimopoulos, 1990 **Low Temperature Physics and Chemistry**, 1982 General Chemistry Brady, 1990-01-02
Modern Chemistry Nicholas D. Tzimopoulos, 1993 **Holt Chemistry** Ralph Thomas Myers, 2004 **Chemistry**
 Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 New York School
Journal, 1895 **Chemical Abstracts**, 1913 **Basic Chemistry** Steven S Zumdahl, 1996

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