

Gas Laws Unit



Diffusion vs. Effusion



- **Diffusion:** gas molecules moving from areas of high concentration to areas of low concentration (mixing of gases)

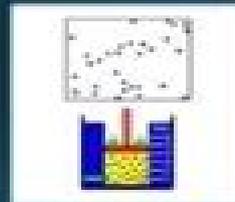
Ideal Gas Law

Rearrange that equation and you get.....THE IDEAL GAS LAW:

$$PV = nRT$$

P = pressure
V = volume
n = moles

Properties of Gases



- They are fluids!
- They have a low density!
- They are highly compressible!
- They completely fill the container they are in!

Unit Bundle



Chemistry Gas Laws Unit Packet

Ensheng Dong



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