

**Honors Chemistry**  
**Chapter 7 Exam Review Sheet**

- Make sure you have read the entire chapter.
- Be able to define all of the chapter 7 vocabulary words.
  - Including:
    - electron dot
    - structure
    - ionic compound
    - valence electron
    - ionic bond
    - chemical formula
    - halide ion
  - alloy
  - octet rule
  - formula unit
  - coordination number
  - metallic bond
- Know the number of valence electrons (electrons in the highest occupied energy level of an atom) for the elements in the representative groups 1A-7A.
- Know the charges of the ions for the elements in the representative groups 1A-7A.
- The maximum charge an ion is likely to have is -3 or +3.
- Be able to write the electron configurations for both the ions and the atoms in the representative groups 1A-7A and for the transition metals.
  - Note: the ions for these elements will have a pseudo noble gas configuration.
- Be able to identify the number of electrons an element gives up or takes on when it forms an ion.
  - Metals donate electrons when they form ions. They will have positive charges. (Cations)
  - Nonmetals take on electrons when they form ions. They will have negative charges. (Anions).
- A pseudo-noble gas for a transition metal contains a **full  $3d$  sublevel**.
  - $Cu^+$ ,  $Zn^{2+}$ , have a pseudo-noble gas configuration.
- Be able to write the correct formula unit for an ionic compound.
  - The overall charge of a **correct formula unit is zero**.
  - use the criss-cross method to write formulas.
- Be able to write the correct electron configuration for all of the elements, all of the ions that the elements form and be able to identify the correct charge for all of the ions that they form.
  - Including but not limited to:**
    - Li, Mg, He, Sn, Si, Ca, Ga, Sr, Ag, Ba, Cu, Fe, Ni, K, Mn, Br, I, Cd, N, P, F, O, Cr,  $Sn^{2+}$ ,  $F^-$ ,  $Sn^{4+}$ ,  $O^{2-}$ .
- The formula for sodium sulfate is  $Na_2SO_4$ .
- The **condensed (abbreviated) electron configuration** distinguishes the core electrons from the valence electrons. **CORE** electrons are tightly held to the nucleus and resemble a noble gas configuration. **VALENCE** electrons are the outer most electrons and are involved in chemical reactions.
- Examples of full & condensed (abbreviated) electron configuration:
 

<u>Full electron configuration</u>	<u>Condensed (abbreviated) electron configuration</u>
a. Li: $1s^2 2s^2$	Li: [He] $2s^2$
b. Co: $1s^2 2s^2 2p^6 3s^2 3p^4 4s^1 3d^7$	Co: [Ar] $4s^1 3d^7$
c. Br: $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10} 4p^5$	Br: [Ar] $4s^2 3d^{10} 4p^5$
d. $S^{2-}$ : $1s^2 2s^2 2p^6 3s^2 3p^4$	$S^{2-}$ : [Ne] $3s^2 3p^4$
e. $Fe^{2+}$ : $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^6$	$Fe^{2+}$ : [Ar] $4s^2 3d^6$
- Alloys a mixture composed of two or more elements, at least one of which is a metal. Their properties are often superior to their component elements. EX:
  - Bronze is tougher than pure copper
  - Steel resists corrosion & is stronger than iron. Steels have a wide range of uses.
  - Some are substitutional alloys and others are interstitial alloys.
- Be able to identify the properties of ionic compounds. Ex:
  - They are made of cations & anions bonded together.
  - They are crystalline solids at room temperature
  - They have high melting points
  - They conduct electricity when they are dissolved in water or melted.
- Be able to identify the shape & coordination number of ions in a crystal. The **coordination number** of an ion is the number of ions of opposite charge that surround the ion in a crystal. Ex:
  - Body-centered cubic, 8
  - Face-centered cubic, 12
  - Hexagonal close-packing, 12

# Chemistry Chapter Test Review

**Thomas A. Evangelist**



## **Chemistry Chapter Test Review:**

DAT Prep Plus 2023-2024 Kaplan Test Prep, 2023-02-07 Kaplan's DAT Prep Plus 2023-2024 provides the test taking strategies, realistic practice, and expert guidance you need to score higher on the Dental Admissions Test. Our comprehensive subject review reflects recent changes to the blueprint of the exam, question types, and test interface. You'll get two full-length practice DATs and expert tips to help you face Test Day with confidence. We're so confident that DAT Prep Plus offers all the knowledge you need to excel at the DAT that we guarantee it: after studying with our online resources and book, you'll score higher on the DAT or you'll get your money back. The Best Review: Two updated full-length online practice exams for test-like practice; Study planning guidance; More than 600 practice questions for every subject with detailed answers and explanations; 12-page full-color study sheets for high-yield review on the go; A guide to the current DAT Blueprint so you know exactly what to expect on Test Day; Comprehensive review of all of the content covered on the DAT; Expert Guidance. Our books and practice questions are written by veteran teachers who know students; every explanation is written to help you learn. Kaplan's experts ensure our practice questions and study materials are true to the test. We invented test prep. Kaplan, www.kaptest.com, has been helping students for 80 years, and our proven strategies have helped legions of students achieve their dreams. Publisher's Note: Products purchased from 3rd party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entities included with the product. Admission Assessment Exam Review E-Book HESI, 2016-02-04. Passing your admission assessment exam is the first step on the journey to becoming a successful health professional. Make sure you're prepared with Admission Assessment Exam Review, 4th Edition. From the testing experts at HESI, this user-friendly guide walks you through the topics and question types found on admission exams, including math, reading comprehension, vocabulary, grammar, biology, chemistry, anatomy, and physiology, and physics. The guide includes hundreds of sample questions, step-by-step explanations, illustrations, and comprehensive practice exams to help you review the subject areas and hone your test-taking skills. Plus, the pre-test and post-test help identify your specific areas of weakness so you can focus your study time on the subjects you need most. If you want to pass the HESI Admission Assessment Exam or any other admissions assessment exam for health professions that you may encounter, there's no better resource than HESI's Admission Assessment Exam Review. HESI Hints boxes offer valuable test-taking tips, as well as rationales, suggestions, examples, and reminders for specific topics. Step-by-step explanations and sample problems in the math section show you how to work through each problem so you understand the steps it takes to complete the equation. Sample questions in all other sections prepare you for the questions you will face on the A2 Exam. User-friendly vocabulary chapter covers more of the medical terminology that you will face on the A2 Exam. Easy-to-read format with consistent section features includes an introduction, key terms, chapter outline, and a bulleted summary to better help you organize your review time and understand the information. Full-color layout and illustrations visually reinforce key concepts for better understanding. NEW 25 question pre

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