

1. How many of the following compounds are named correctly?

I. Na_2SO_4 sodium sulfate ✓
 II. CaCO_3 calcium carbonate ✓
 III. PCl_5 phosphorus pentachloride ✓
 IV. CoO cobalt(II) oxide ✓
 (A) 4
 (B) 3
 (C) 2
 (D) 1

2. Which of the following chemical formulas is incorrect?

A. FeS iron(II) sulfide
 B. SF_6 sulfur hexafluoride ✓
 C. PbO_2 lead(IV) oxide
 (D) NH_4SO_4 ammonium sulfate

3. Which of the following compounds is named incorrectly?

A. PCl_3 phosphorus trichloride ✓
 (B) MgCl_2 magnesium dichloride
 C. Cu_2O copper(II) oxide ✓
 D. At least two of the above (A–C) are named incorrectly.

4. Which of the following names is incorrect?

A. cobalt(II) chloride
 B. magnesium oxide ✓
 (C) aluminum(III) oxide
 D. diphosphorus pentoxide

5. In which case is the polyatomic ion named incorrectly?

(A) NH_4^+ ammonium ✓
 (B) SO_4^{2-} sulfate ✓
 C. CO_3^{2-} carbonate ✓
 D. NO_2^- nitric ✓

6. What is the correct name for the compound Fe_2O_3 ?

(A) iron oxide
 (B) iron(III) oxide ✓
 C. iron(II) oxide
 D. diiron trioxide

(i.e. (III) = oxide)

7. Which of the following is the correct formula for chlorous acid?

A. HClO
 (B) HClO_2
 C. HClO_3
 D. HClO_4

8. When dihydrogen sulfide is dissolved in water, it is called

A. sulfuric acid.
 (B) sulfurous acid. H_2S
 C. hydrochloric acid.
 D. dihydrochloric acid.

9. Which of the following acids is named incorrectly?

(A) HNO_3 nitrous acid
 B. HCl hydrochloric acid ✓
 C. $\text{HC}_2\text{H}_3\text{O}_2$ acetic acid ✓
 D. H_3PO_4 phosphoric acid ✓

10. Fill in the missing name or formula.

Name	Formula
barium phosphate	$\text{Ba}_3(\text{PO}_4)_2$
dinitrogen monoxide	N_2O
sulfuric acid	H_2SO_4
copper (I) oxide	Cu_2O
boron trifluoride	BF_3

11. In naming binary compounds, you sometimes simply name the ions. In other cases, you need to use Roman numerals or prefixes.

- Explain why Roman numerals are sometimes needed in naming binary compounds, and provide an example name and formula.
- Explain why prefixes are sometimes needed in naming binary compounds, and provide an example name and formula.

a) when naming binary compounds, Roman numerals are needed when a cation has more than one possible charge. (i.e. FeS → iron(II) sulfide)

b) prefixes are needed when naming covalent compounds (2 nonmetals)
 (PCl_5 → phosphorus pentachloride)

Chemistry Chapter Standardized Test Practice Answers

Huangqi Zhang



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