

Chemistry Chapter 6 Review **KEY**

1. What is the difference between a covalent bond and an ionic bond?
Covalent – share electrons to reach a full octet (Example: H—O—H)
ionic – steals electrons to reach a full octet (Example: Na⁺ Cl⁻)

2. Determine whether the bond formed between each pair of atoms will be ionic or covalent:
 - a. K and I **ionic**
 - b. Mg and O **ionic**
 - c. C and Cl **covalent**
 - d. Ca and N **ionic**
 - e. Li and O **ionic**
 - f. C and H **covalent**

3. Then write the formula of the ionic compound. (Remember that the positive and negative charges must cancel out.)
 - a. Sodium and oxygen Na⁺ O²⁻ **Na₂O**
 - b. Magnesium and chlorine Mg²⁺ Cl⁻ **MgCl₂**
 - c. Potassium and fluorine K⁺ F⁻ **KF**
 - d. Calcium and carbonate (CO₃²⁻) Ca²⁺ CO₃²⁻ **CaCO₃**
 (Note: This is a polyatomic ion, meaning that it is a covalent compound with a charge. You will become experts on them in Unit 7.)

4. In your own words, define **polarity**:
One atom pulls more strongly on the atoms in a bond. Unequal distribution of electrons in a bond. The one Border collie pulls more strongly on the toy than the other Border collie, but instead of border collies they are atoms and instead of a chew toy it is electrons. So on and so forth.

5. For each bond listed below, label the polarity by drawing an arrow in the correct direction (pointing towards the more electronegative atom).
 - a. $\overset{\text{S}}{\text{---}} \overset{\text{O}}{\text{---}} \rightarrow$
 - b. $\overset{\text{H}}{\text{---}} \overset{\text{Cl}}{\text{---}} \rightarrow$
 - c. $\overset{\text{C}}{\text{---}} \overset{\text{F}}{\text{---}} \rightarrow$

Chemistry Chapter 6 Answer Key File

Rachel S Tattersall



Chemistry Chapter 6 Answer Key File:

Holt Chemistry File, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

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boundary value problems of the 1D reaction diffusion equations on any finite interval of the real line *Radioactive Waste Management*, 1981

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