

AQA Chemistry

GCSE Exam-style mark scheme

C1 Atomic structure

9099 Combined Science: Trilogy

11/2019

| Question | Answers | Extra Information | Mark | AO / Spec ref. |
|----------|--|---|-----------------------|------------------------------------|
| 01.1 | $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ | Allow correct multiples; allow 1 mark for Na and NaCl. | 2 | AO2 C1.1.1 |
| 01.2 | <p>Level 3 (5–6 marks): A full, clear description of the process is provided. The major steps are outlined in a logical method that would result in the required products being obtained.</p> <p>Level 2 (3–4 marks): The substantive content of the process is present, but some steps are missing. The explanation may not be in a completely logical sequence.</p> <p>Level 1 (1–2 marks): Simple statements relating to the process are made but with no logical structure.</p> <p>Level 0 (0 marks): No relevant content.</p> <p>Indicative Content</p> <ul style="list-style-type: none"> • add water • stir/warm mixture • filter to remove sand • sand washed to remove salt solution • sand dried in warm oven • filtrate/salt solution heated • suitable method for crystallisation, e.g., salt solution heated in evaporating basin on a water bath until crystals first appear on side of dish or half volume of solution evaporated • leave to crystallise in warm place or leave for a few days. <p>This indicative content is not exhaustive, other creditworthy responses should be awarded marks as appropriate.</p> | | 6 | AO1 C1.1.2 |
| 02 | 4 protons; 5 neutrons; 4 electrons; protons and neutrons in nucleus; electrons in energy levels/shells. | Allow electronic configuration is 2,2. | 1 1 1 1 1 | AO1×3 AO2×2 C1.1.4 C1.1.7 |
| 03.1 | CO_2 | | 1 | AO3 C1.1.1 |

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