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<p>15 <b>P</b> Phosphorus 30.9738 <math>[\text{Ne}]3s^2 3p^3</math></p>	<p>16 <b>S</b> Sulfur 32.066 <math>[\text{Ne}]3s^2 3p^4</math></p>	<p>17 <b>Cl</b> Chlorine 35.453 <math>[\text{Ne}]3s^2 3p^5</math></p>
<p>33 <b>As</b> Arsenic 74.9216 <math>[\text{Ar}]3d^{10} 4s^2 4p^3</math></p>	<p>34 <b>Se</b> Selenium 78.96 <math>[\text{Ar}]3d^{10} 4s^2 4p^4</math></p>	<p>35 <b>Br</b> Bromine 79.904 <math>[\text{Ar}]3d^{10} 4s^2 4p^5</math></p>

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# Chemistry Chapter 12 Reviewing Content Answers

**Michael Cox, Aaron Hoskins, Alain  
Viel, Judith Simcox**



## Chemistry Chapter 12 Reviewing Content Answers:

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