

# Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- What type of charge does a proton have? \_\_\_\_\_
- What type of charge does a neutron have? \_\_\_\_\_
- What type of charge does an electron have? \_\_\_\_\_
- Which two subatomic particles are located in the nucleus of an atom? \_\_\_\_\_

2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.

8	← Atomic Number
O	← Symbol
Oxygen	← Name
15.999	← Average atomic mass

6	#P _____
C	#N _____
Carbon	#E _____
12.011	

10	#P _____
Ne	#N _____
Neon	#E _____
20.108	

19	#P _____
K	#N _____
Potassium	#E _____
39.098	

3. Complete the table using your knowledge of the periodic table.

Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na			11		12
K		39		19	
			38		50
F				9	10
	20	40		20	
	50			50	69
I	53	127			

# Chemistry Basic Atomic Structure Worksheet Answers Chart

**Carolina Biological Supply Company**



## **Chemistry Basic Atomic Structure Worksheet Answers Chart:**

**Simplified ICSE Chemistry** Viraf J. Dalal, **Learning Elementary Chemistry for Class 8 (A.Y. 2023-24)** Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series investigations related to the text enable the learners to learn through experimentation Quick revision of each chapter has been given under the caption Highlights in Review Online Support It provides Video lectures Unit wise interactive exercises Chapterwise Worksheet Solution of textbook questions for Teachers only E Book for Teachers only I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book Author **Biology/science Materials** Carolina Biological Supply Company, 1991 **Excel Basics to Blackbelt** Elliot Bendoly, 2013-07-29 This second edition is designed for those interested in developing their skills in the construction of analytical tools and rich visual demonstrations in Excel *Problems and Problem Solving in Chemistry Education* Georgios Tsaparlis, 2021 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative

reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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