

Reinforcement: The Chemistry of Life

elements
enzymes
neutrons
macromolecules

periodic
protons
compounds
covalent

organic
electrons
cohesion
capacity

energy
ionic
adhesion
organic

hydrogen
polarity
van der Waals
nucleus



- Neutral subatomic particles found in the nucleus of an atom: _____
- Positively charged subatomic particles that make up atoms: _____
- Negatively charged subatomic particles that make up atoms: _____
- Type of bond that involves transferring electrons: _____
- Type of bond that involves sharing electrons: _____
- Substances that consist of only one type of atom: _____
- Substances that consist of two or more elements: _____
- Weak interactions responsible for the "sticky feet" of geckos: _____ forces.
- Lactase is an _____ that breaks down milk sugar.
- Molecules that make up living things and contain carbon are called _____ molecules.
- Large molecules made of many small subunits: _____
- Type of bond that causes water to stick to other water molecules: _____
- Type of bond that causes water to stick to other things: _____
- Compounds that consist of chains of carbon: _____
- A large amount of energy is needed to change water temperature. It has a high heat _____
- Show elements arranged in order by atomic number. Use _____ table.
- _____ refers to molecules that have a partial negative and positive charge.
- Formed between water molecules, type of weak bond: _____ bond.
- The center of an atom, contains protons and neutrons: _____
- Changing the temperature of water from cold to warm requires _____
- Sketch a molecule of water and indicate which areas are positive and which areas are negative.
- Explain how a water spider stays on the surface of water.

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Chapter 2 Reinforcement How Elements Bond Answer Key

Ian Pickup



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