

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 9

## STUDY GUIDE FOR CONTENT MASTERY

## Covalent Bonding

### Section 11 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond
---------------	----------	------------	------------	---------

When sharing electron pairs, the attachment between atoms that results is called a(n) **covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **pi bond**.

In your textbook, read about single and multiple bonds and bond strength.

Circle the letter of the choice that best completes the statement or answers the question.

- Is a pair from an element such as hydrogen, oxygen, and nitrogen normally covalent?
  - an ionic atom
  - an molecule containing three atoms
  - an molecule containing two atoms
  - an molecule containing four atoms
- How many electrons are shared in a double covalent bond?
  - one
  - two
  - three
  - four
- Bond length is the distance between
  - two molecules of the same substance
  - the nucleus of two attached atoms
  - the electron in two attached atoms
  - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
  - As the electron bond, the stronger the bond.
  - As the electron bond, the weaker the bond.
  - As the electron bond, the lower the electron is a.
  - As the electron bond, the lower the bond dissociation energy.

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 9



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## CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

### Section 12 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write true or false.

- false** 1. Binary molecular compounds are generally composed of a metal and a nonmetal.
- false** 2. The second element in the formula of a binary compound is named using the suffix -ite.
- false** 3. The prefix zero indicates three atoms.
- true** 4. The prefix four indicates six atoms.
- true** 5. In naming the first element in a formula, the prefix zero is not used.
- true** 6. For binary acids, the hydrogen part of the compound is named using the prefix hydro.
- false** 7. An oxyacid contains only two elements.
- false** 8. If the name of the cation of an ionic salt is not, the salt name contains the suffix -ate.

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

## Column A

- c 1.  $\text{SO}_2$
- i 2.  $\text{CO}$
- g 3.  $\text{H}_2\text{CO}_3$
- f 4.  $\text{H}_2\text{SO}_4$
- b 5.  $\text{N}_2\text{O}_5$
- d 6.  $\text{HNO}_3$
- i 7.  $\text{HNO}_2$
- a 8.  $\text{HBr}$
- h 9.  $\text{HBrO}_3$

## Column B

- a.** hydrobromic acid
- b.** nitrogen trioxide
- c.** carbonic acid
- d.** nitric acid
- e.** carbonic acid
- f.** nitrous acid
- g.** carbonic acid
- h.** bromic acid
- i.** carbon dioxide

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# Chapter 1 study Guide For Content Mastery Answer Key

**Jin-Ying Zhang**



**Chapter 1 study Guide For Content Mastery Answer Key:**

## **Chapter 1study Guide For Content Mastery Answer Key** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 1study Guide For Content Mastery Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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