

Lewis Structure Worksheet

1. Draw the Lewis Structure for each of the following elements.

Element	Lewis Structure	Element	Lewis Structure	Element	Lewis Structure
Lithium	Li	Nitrogen	N	Silicon	Si
Beryllium	Be	Fluorine	F	Neon	Ne

2. Draw the Lewis Dot Diagram for each of the following elements. Determine how many electrons the atom has to gain or lose to become stable (fill or lose outer orbit to make 8). Determine the charge that the ion will have (recall: gain electrons = "-" charge, lose electrons = "+" charge).

Element	Lewis Structure	Needed for Stability	Charge on ion
Sodium	Na	Lose 1	+1
Oxygen	O	Gain 2	-2
Aluminum	Al	Lose 3	+3
Helium	He	N/A	N/A
Potassium	K	Lose 1	+1
Carbon	C	4	Does not form an ion

Chemistry Worksheets Lewis Dot Structures Answer Key

Bowker Editorial Staff



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