

# Chemistry Study Guide

Periodic Table of the Elements

|    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |
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| H  | He |    |    |    |    |    |    |    |    |    |    | Li | Be | B  | C  | N  | O  | F  | Ne |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |
| Na | Mg |    |    |    |    |    |    |    |    |    |    | Al | Si | P  | S  | Cl | Ar |    |    |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |
| K  | Ca | Sc | Ti | V  | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr |    |    |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |
| Rb | Sr | Y  | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In | Sn | Sb | Te | I  | Xe |    |    |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |
| Cs | Ba |    |    |    |    |    |    |    |    |    |    | Hf | Ta | W  | Re | Os | Ir | Pt | Au | Hg | Tl  | Pb | Bi  | Po | At  | Rn  |    |  |  |  |  |  |  |  |  |  |
| Fr | Ra |    |    |    |    |    |    |    |    |    |    | Rf | Db | Sg | Bh | Hs | Mt | Ds | Rg | Cn | Uut | Fl | Uup | Lv | Uus | Uuo |    |  |  |  |  |  |  |  |  |  |
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|    |    | Ac | Th | Pa | U  | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No | Lr |    |    |    |    |     |    |     |    |     |     |    |  |  |  |  |  |  |  |  |  |

## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

## Matter:

Anything that has **mass** and **occupies space** ✨🌍

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location    |
|----------|--------|------------|--------|-------------|
| Proton   | $p^+$  | 1          | +1     | Nucleus     |
| Neutron  | $n$    | 1          | 0      | Nucleus     |
| Electron | $e^-$  | $\sim 0$   | -1     | Orbit/Shell |

# Chemistry Study Guide Spring 2013

**Christian Drosten**



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