

Chemistry Study Guide

Periodic Table of the Elements

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| 3 | 4 | | | | | | | | | | | Na | Mg | Al | Si | P | S | Cl | Ar | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 | 26 | 27 | 28 | 29 | 30 | 31 | 32 | 33 | 34 | 35 | 36 | 37 | 38 | 39 | 40 | 41 | 42 | 43 | 44 | 45 | 46 | 47 | 48 | 49 | 50 | 51 | 52 | 53 | 54 | 55 | 56 | 57 | 58 | 59 | 60 | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 | 71 | 72 | 73 | 74 | 75 | 76 | 77 | 78 | 79 | 80 | 81 | 82 | 83 | 84 | 85 | 86 | 87 | 88 | 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 | 104 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 | 113 | 114 | 115 | 116 | 117 | 118 |
| K | Ca | Sc | Ti | V | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr | Rb | Sr | Y | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In | Sn | Sb | Te | I | Xe | Cs | Ba | | | | | | | | | | | Rn | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
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| | | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb | Lu | Ac | Th | Pa | U | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No | Lr | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|----------|--------|------------------|--------|-------------|
| Proton | p^+ | 1 | +1 | Nucleus |
| Neutron | n | 1 | 0 | Nucleus |
| Electron | e^- | $\frac{1}{1836}$ | -1 | Orbit/Shell |

Chemistry Study Guide Answers Unit 7

J Rink



Chemistry Study Guide Answers Unit 7:

The book delves into Chemistry Study Guide Answers Unit 7. Chemistry Study Guide Answers Unit 7 is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry Study Guide Answers Unit 7, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemistry Study Guide Answers Unit 7
- Chapter 2: Essential Elements of Chemistry Study Guide Answers Unit 7
- Chapter 3: Chemistry Study Guide Answers Unit 7 in Everyday Life
- Chapter 4: Chemistry Study Guide Answers Unit 7 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chemistry Study Guide Answers Unit 7. The first chapter will explore what Chemistry Study Guide Answers Unit 7 is, why Chemistry Study Guide Answers Unit 7 is vital, and how to effectively learn about Chemistry Study Guide Answers Unit 7.

3. In chapter 2, the author will delve into the foundational concepts of Chemistry Study Guide Answers Unit 7. This chapter will elucidate the essential principles that must be understood to grasp Chemistry Study Guide Answers Unit 7 in its entirety.

4. In chapter 3, this book will examine the practical applications of Chemistry Study Guide Answers Unit 7 in daily life. The third chapter will showcase real-world examples of how Chemistry Study Guide Answers Unit 7 can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Chemistry Study Guide Answers Unit 7 in specific contexts. This chapter will explore how Chemistry Study Guide Answers Unit 7 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Chemistry Study Guide Answers Unit 7. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Study Guide Answers Unit 7.

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